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# Orbital distortion and electric field control of sliding ferroelectricity in a boron nitride bilayer

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## Abstract

Recent experiments confirm that two-dimensional (2D) boron nitride (BN) films possess room-temperature out-of-plane ferroelectricity when each BN layer is sliding with respect to each other. This ferroelectricity is attributed to the interlayered orbital hybridization or interlayer charge transfer in previous work. In this work, we attempt to understand the sliding ferroelectricity from the perspective of orbital distortion of long-pair electrons. Using the maximally localized Wannier function (MLWF) method and first-principles calculations, the out-of-plane  $p_z$  orbitals of BN are investigated. Our results indicate that the interlayer van der Waals interaction causes the distortion of the N  $p_z$  orbitals. Based on the picture of out-of-plane orbital distortion, we propose a possible mechanism to tune the ferroelectric polarization by external fields, including electric field and stress field. It is found that both the polarization intensity and direction can be modulated under the electric field. The polarization intensity of the system can also be controlled by stress field perpendicular to the plane. This study will provide theoretical help in the device design based on sliding ferroelectrics.

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## 1. Introduction

Ferroelectric materials possessing nonvolatile and switchable spontaneous electric polarization are essential elements in diverse technology applications, including memories, field-effect transistors, solar cells, sensors, and actuators. Over the past decade, two-dimensional (2D) ferroelectric materials have attracted extensive attention due to their potential applications in the field of microelectronics. Compared with traditional ferroelectric materials, 2D ferroelectric materials are more easily integrated into functional components in miniaturized electronic devices because of their inherent nano size and ferroelectric properties [1]. In the past few years, with the large number of theoretical predictions of 2D ferroelectric materials [2-7], the intriguing ferroelectric properties of many 2D materials down to atomic thickness have been experimentally studied [8-18]. Recently, Yasuda et al. and Vizner Stern et al. showed that ferroelectricity can be engineered by artificially stacking a 2D nonpolar material such as boron nitride (BN) [19,20]. Authors attribute the sliding ferroelectricity to the interlayer orbital hybridization. Theoretical studies proposed that the ferroelectric polarization originated from net charge transfer from the upper layer to the down layer [5]. In this work, we propose the origination of sliding ferroelectricity may be understood from the orbital distortion induced by the interlayer coupling. Moreover, the tunability of the sliding ferroelectricity by external field still lack investigations in both experiments and theory, and therefore need to be addressed.

To control the ferroelectric properties of 2D materials, many strategies have been proposed, such as atomic substitution, chemical modification, supporting substrates, and application of external electric fields. For ferroelectric device, the controlling of external fields on polarization is undoubtedly more meaningful than other tuning methods. In fact, the transport, magnetism, catalysis, and other properties of 2D materials can all be modified by an external field [21-36]. A large number of relevant studies have been reported [23,33-39]. However, the external electric field effects on

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4 the sliding ferroelectricity of 2D materials are rarely reported.  
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7 In this work, we first discuss the role of orbital distortion on the sliding  
8 ferroelectricity, and then study the ferroelectric polarization switching induced by  
9 external field application to 2D BN. We use the first-principles calculation method and  
10 the maximally localized Wannier function (MLWF) method [40] to confirm that the  
11 electric polarization comes from the distortion of the B-top N (N atom facing a B atom)  
12  $2p_z$  orbital, but the distortion of the hexagon-top N (N atom facing empty centre of the  
13 hexagon)  $2p_z$  orbital also plays an important role. There is a linearly varying  
14 relationship between the polarization and electric field. Interestingly, with an increase  
15 in the electric field strength, only the distorted shape of the  $p_z$  orbital of the hexagon-  
16 top N atom greatly changes, while the distorted shape of the  $p_z$  orbital of the B-top N  
17 atom changes slightly. This is due to the different environments of the N atoms in  
18 different layers. For the N1 atom, the N  $2p_z$  orbital is not only attracted by the  $B^{3+}$  ion  
19 but also repulsed by the B  $2p_z$  orbital. For the N2 atom, the N  $2p_z$  orbital is only  
20 subjected to the electrostatic repulsion of the  $p_z$ -orbital on the hexagon. Additionally,  
21 we find that, external pressure helps drive stronger interlayer interactions and enhances  
22 the polarization of AB (BA) stacked bilayer BN. Furthermore, the MLWF method has  
23 been used for studies of various properties of materials but rarely for studies of  
24 ferroelectricity [41-45]. This work hence bridges knowledge from different fields.  
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## 42 **2. Computational methods**

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45 In this work, all calculations are performed using the density-functional theory  
46 (DFT) method, which is implemented in the Vienna Ab initio Simulation Package  
47 (VASP) [46-48] within the projector augmented wave (PAW) method [49]. The Perdew,  
48 Burke, and Ernzerhof (PBE) [50] parameterized generalized gradient approximation  
49 (GGA) is chosen to describe the exchange-correlation functional. To accurately describe  
50 the vdW interaction in the bilayer BN systems, we adopt the empirical correction  
51 method presented by Grimme (DFT-D3) [51]. The climbing image nudged elastic band  
52 (CI-NEB) method is used to calculate the energy barriers of transition states [52].  
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4 MLWF [40,53,54] method is employed to calculate the spatial distribution of N  $p_z$   
5 orbitals. The cut-off energy for the basis set is chosen to be 500 eV. All geometric  
6 structures are fully relaxed until the force on each atom is less than 0.02 eV/Å, and the  
7 convergence criterion is  $10^{-6}$  eV for the electronic energy. The atomic structure is  
8 studied with a periodic slab geometry, adopting a sufficiently large vacuum layer of 15  
9 Å to avoid interactions between adjacent layers. The optimized lattice parameter of the  
10 AB/BA stacking BN bilayer is 2.50 Å, which is in good agreement with previous values  
11 [55-57]. For geometry optimization,  $9 \times 9 \times 1$  special  $k$ -points are used for sampling  
12 the Brillouin zone, and denser  $12 \times 12 \times 1$  special  $k$ -points are set for electronic self-  
13 consistent calculations.  
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### 26 **3. Results and discussion**

#### 27 **3.1. Origin of ferroelectricity in AB (BA) stacked bilayer BN**

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31 Bulk  $h$ -BN crystals realize AA' stacking, as shown in reference [56]. This natural  
32 stacking order of a  $180^\circ$ -rotation restores the broken inversion symmetry in the  
33 monolayer. However, theory and experiments show that if two BN monolayers are not  
34 rotated when stacked (stacked in parallel), a polar AB or BA stacking order will be  
35 formed [56,58-60]. These configurations are parallel stacked forms with minimal local  
36 energy and are implemented as metastable crystal structures [56,58]. In AB (BA)  
37 stacking, the upper layer B (N) atom is above the lower layer N (B) atom, while the  
38 upper layer N (B) atom is above the hexagon center of the lower layer.  
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48 Let us now discuss the origin of the ferroelectric polarization. Initially, we thought  
49 that the ferroelectricity originated from hybridization between the vertically aligned B  
50  $p_z$  orbital and N  $p_z$  orbital, which distorted the N  $2p_z$  orbital. However, as seen from our  
51 calculated projected energy band structure (see figure. 1(a)) and projected density of  
52 states (see figure. 1(b)), there is no obvious hybridization between the vertically aligned  
53 B  $p_z$  orbital and N  $p_z$  orbital. Moreover, the charge transfer between the bilayer BN  
54 layers is found to be negligible by Bader charge analysis [61,62], leaving only vdW  
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4 interactions between the layers, which makes the orbital hybridization and charge  
5 transfer mechanisms invalid for the ferroelectric polarization. Previously, one of our  
6 listed authors found that the dimensional effect of SnTe ferroelectricity can be attributed  
7 to enhancement of Te  $p_z$  electron localization [63]. Thus what is the effect of N  $p_z$   
8 electrons on the bilayer BN ferroelectricity?  
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14 As shown in figure. 1(c), (d) and (e), we calculate the N  $p_z$  orbitals of monolayer  
15 and bilayer BN using the MLWF method. According to the calculation results, the N  $p_z$   
16 orbitals in monolayer BN are symmetric dumbbell-shaped with opposite phase signs of  
17 wannier function. However, the N  $p_z$  orbitals in bilayer AB or BA stacked BN become  
18 asymmetric or distorted. figure. 1(d) shows the unit cell of AB-stacked bilayer BN. The  
19  $p_z$  orbitals of the two N atoms in a unit cell are distorted. The distortion of the  $p_z$  orbitals  
20 shifts the charge center upwards, creating a net dipole. Hence, the polarization direction  
21 of the AB stacking is along the negative  $z$ -axis ( $-P$ ). The opposite case is valid for the  
22 BA stacking, whose polarization is along the positive  $z$ -axis ( $+P$ ) (see figure. 1(e)).  
23 From the above results, the ferroelectric polarization of AB- or BA stacked bilayer BN  
24 originates from distortion of the  $p_z$  orbitals of all N atoms.  
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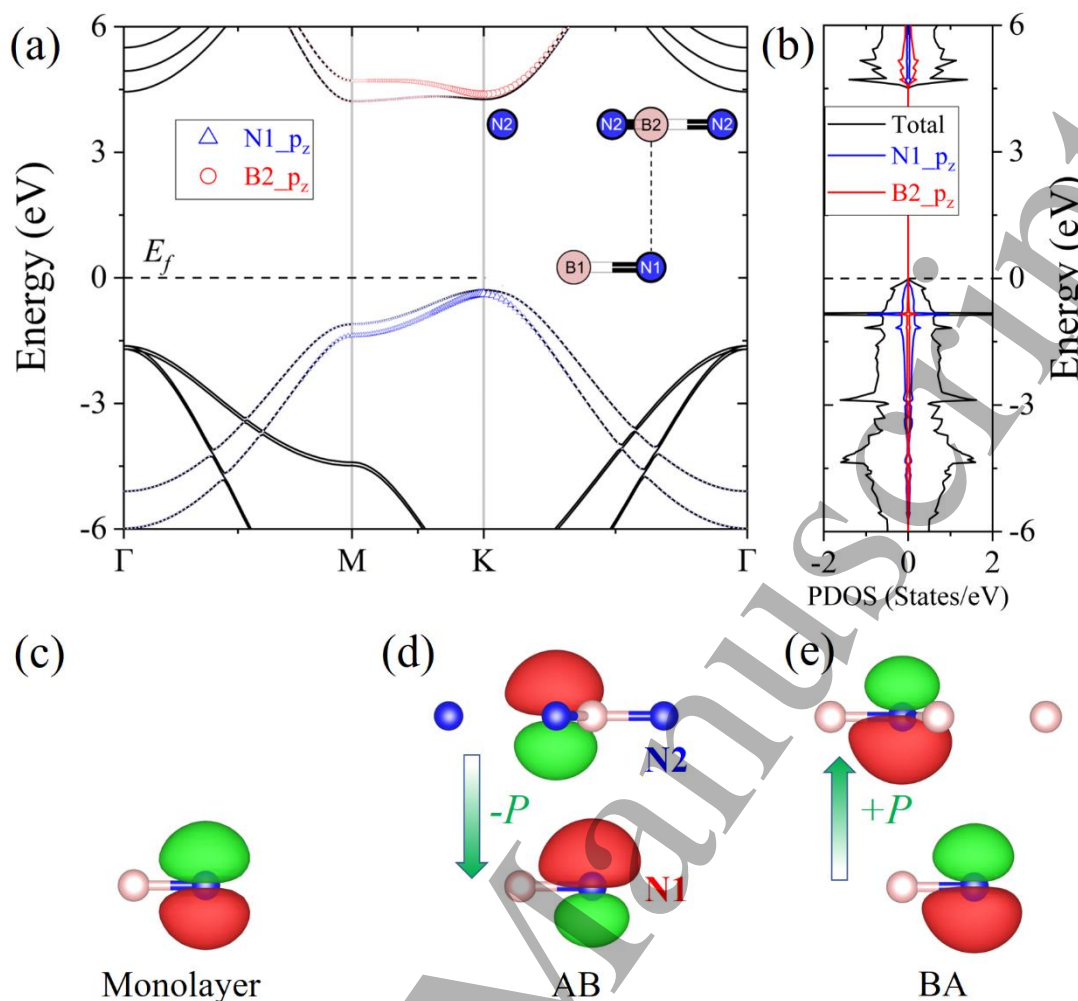


Figure. 1. The projected energy band structure (a) and projected density of states (b) of vertically aligned B atoms and N atoms as shown in the inset. The Fermi energy ( $E_f$ ) is set to zero. (c), (d) and (e) represent the calculated Wannier functions of N  $p_z$  orbitals for monolayer BN, AB and BA stacking BN bilayer, where red (+) and green (-) represent the phase sign of the Wannier function.

The N atoms in different layers are in different chemical environments, which leads to different degrees of distortion of their  $p_z$  orbitals. The situation of BA stacking is completely opposite to that of AB stacking, and we will only discuss the structure of AB stacking here. In the AB stacking, the B-top N atoms (N1) lie beneath the top layer B atoms. There are both attraction and repulsion between these vertically aligned N atoms and B atoms (see figure. 2(a) and (b)). The repulsion originates from N  $p_z$  electrons and B  $p_z$  electrons (see figure. 2(a)), which can be seen according to the

momentum path interval of  $\Gamma \leftrightarrow M$  (figure. 2(d)). figure. 2(e) is the partial charge density at point ① in figure. 2(d). The red energy band corresponds to the  $p_z$  orbital of the N1 atom. The partial charge density in the  $\Gamma \leftrightarrow M$  interval on the band is similar to that in figure. 2(e), which is consistent with our model in figure. 2(a). The attraction originates from N  $p_z$  electrons and  $B^{3+}$  ions (see figure. 2(b)), which can be seen according to the band structure in the  $M \leftrightarrow K$  interval (figure. 2(d)).

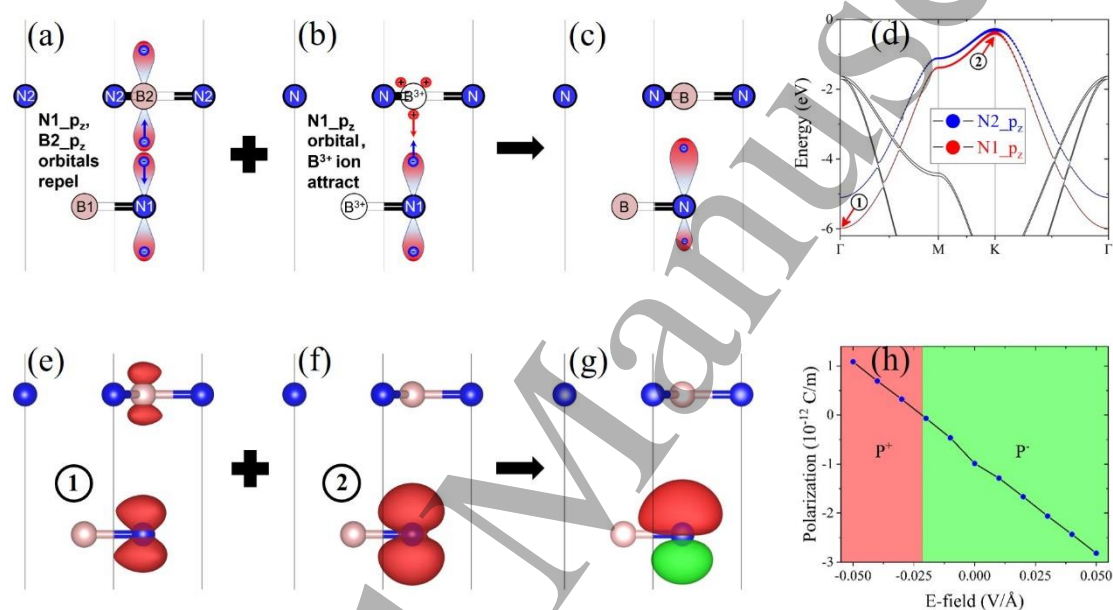


Figure. 2. (a)-(c) Schematic representation of N atom  $p_z$  orbital distortion with B and N atoms arranged vertically in a bilayer AB stacking BN. (c) Schematic diagram of N  $p_z$  orbital after distortion. (d) Projected energy band of AB stacking bilayer BN. The red and blue represents the lower layer N atom and the upper layer N atom respectively. (e) and (f) are the partial charge density at points ① and ② in figure 2(d), respectively. (g) The  $p_z$  orbital of the lower layer N atom calculated by the MLWF method. (h) Dependence of the electric polarization of bilayer AB stacking BN on the external electric field.

Figure. 2 (f) shows the partial charge density at point ② in figure. 2(d). The partial charge density in the  $M \leftrightarrow K$  interval on the N  $p_z$  band of the upper layer is similar to



that in figure. 2(f), which is consistent with our model in figure. 2(b). The competition between attraction and repulsion distorts the N  $p_z$  orbital of the upper layer as shown in figure. 2(c). This is consistent with the N1  $p_z$  orbital as calculated by the MLWF method (see figure. 2(g)). The existence of these two interactions is also the reason for the stable existence of AB (BA) stacked bilayer BN. From figure. 2(d), we can conclude that the N1  $p_z$  orbital energy is lower than that of the hexagon-top N (N2)  $p_z$  orbital. These two interactions also stabilize the distorted N1  $p_z$  orbital. When we apply forward and reverse electric fields (the electric field is perpendicular to the 2D BN plane), the electric field preferentially modulates the shape of the N2  $p_z$  orbital, while the shape of the N1  $p_z$  orbital hardly changes (see figure. 3). Compared with the N1  $p_z$  orbital, the N2  $p_z$  orbital is only slightly distorted by the electrostatic repulsion of the  $p_z$  orbitals of B and N atoms on the facing hexagon (see figure, 1(d)). The distance between the bilayer is  $d = 3.43 \text{ \AA}$ . According to the charge center method, we calculate the vertical polarization of AB stacking to be  $-0.988 \times 10^{-12} \text{ C/m}$ , which is in the same order of magnitude with the Berry phase calculation of  $-0.805 \times 10^{-12} \text{ C/m}$  and previously reported value  $-2.08 \times 10^{-12} \text{ C/m}$  [5].

### 3.2 External electric field effects on the bilayer BN

The external electric field plays a vital role in the modulation of the electronic properties of 2D materials. In the following, we present the effect of an external uniform electric field on the electronic properties of bilayer BN for two different stacking configurations, namely, AB and BA. The electric field is applied perpendicular to the plane of the BN bilayer. The electric field strength is considered to be between 0 and  $0.3 \text{ V/\AA}$ . figure 2(h) shows the relationship between the intrinsic ferroelectric polarization of AB stacked bilayer BN and the external electric field calculated by the charge centre method. When the applied electric field strength is less than  $-0.02 \text{ V/\AA}$ , the intrinsic ferroelectric polarization of the AB stacking is reversed. As shown in figure. 3(c), (d) and (e), the electric field tunes the polarization of AB stacked BN by adjusting the shape of the N2  $p_z$  orbital. This also proves that the distorted N1  $p_z$  orbital is more stable. When the electric field strength is  $0.05 \text{ V/\AA}$ , the polarization strength is  $-2.819$

$\times 10^{-12}$  C/m, and when the electric field strength is  $-0.05$  V/Å, the polarization strength is  $1.088 \times 10^{-12}$  C/m. Therefore, the strength and direction of the intrinsic ferroelectric polarization of BN can be regulated by an external field.

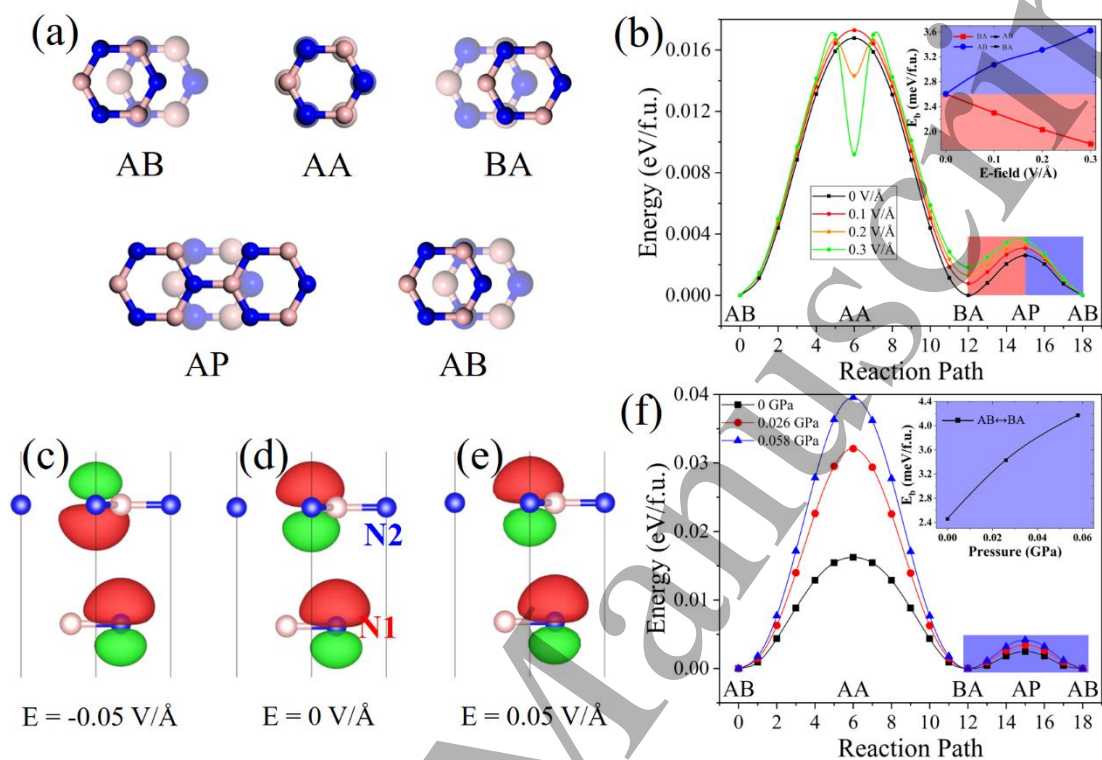


Figure 3. Ferroelectric switching pathway of bilayer BN. (a) The top view of the structure of bilayer BN at the extreme point in figure (b) and (f). For clarity, the top (bottom) layer are represented (dimmed) by smaller (larger) atoms. (c)-(e) Variation of  $p_z$  orbital shape of N atom with applied electric field calculated by the MLWF method. The applied electric fields are  $-0.5$  V/Å (c) and  $0.5$  (e) V/Å, respectively. (d) Pure AB stacked bilayer BN. (b) CI-NEB for the energy pathway for the AB  $\leftrightarrow$  BA stacked bilayer BN transformation in different applied electric fields. The inset represents the AB  $\leftrightarrow$  BA polarization switching energy barrier versus the applied electric field. (f) CI-NEB for the energy pathway for the AB  $\leftrightarrow$  BA stacked bilayer BN transformation in different applied pressure. The inset represents the AB  $\leftrightarrow$  BA polarization switching energy barrier versus the applied pressure.

From the above discussion, we know that the ferroelectric polarization

perpendicular to the bilayer BN plane can be switched by sliding one layer with respect to the other layer. The sliding pathway, *i.e.*, the ferroelectric switching pathway, is computed by using the CI-NEB method, and is shown in figure. 3(b). There are two different sliding pathways between AB and BA stacking, namely, AB→AA→BA (the AB phase slips through the AA phase to the BA phase) and AB→AP→BA (the AB phase slips through the AP phase to the BA phase). According to our calculation results, the sliding path AB→AP→BA has a low energy barrier, and the AA phase is not a stable phase in the absence of an external electric field. When the sliding energy barrier are calculated under the applied electric fields, we find that the unstable AA phase becomes stable when the electric field is 0.2 V/Å. When there is no external electric field or the external electric field is relatively small, the energy of the AA phase is at the saddle point of the energy pathway. When the external electric field strength is greater than 0.2 V/Å, the energy of the AA phase is lower than the saddle point of the energy pathway. This means that the AA, AB, and BA phases are all stable in the presence of an electric field. This result can provide theoretical help for device design. For example, in the absence of an external electric field, only two stable ferroelectric phases AB and BA can exist in the system. In this case the ferroelectric polarization ( $P$ ) is switched only between  $P_{\downarrow}$  and  $P_{\uparrow}$ . When the applied electric field strength is greater than 0.2 V/Å, there will be three stable phases, AA, AB and BA. At this time, the ferroelectric polarization will switch between  $P_0$ ,  $P_{\downarrow}$  and  $P_{\uparrow}$ .

The existence of the additional paraelectric phase AA has many possibilities in the design of logic devices. The inset of figure. 3(b) presents the switching energy barrier between the AB and BA stacking. When there is no external electric field, the sliding energy barrier between the AB and BA stacking is 2.6 meV/fu, which is close to the previous theoretical calculations [5]. When an external field exists, the sliding energy barrier from AB to BA increases, while the sliding energy barrier from BA to AB decreases. Therefore, the sliding energy barrier for AB↔BA can be regulated by the external field.

### 3.3 Pressure effects on the bilayer BN

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4 Altering the electronic structure and interlayer distance by applying high pressure  
5 is a promising strategy to tune the interlayer coupling, which is a well-established  
6 approach for tuning physical properties, but it is rarely used in the study of ultrathin  
7 quantum systems. High pressure has been shown in the past few years to be a powerful  
8 tool to change the electronic structure of materials and even promote the phase  
9 transition of materials [64,65]. The effect of pressure on the ferroelectric properties of  
10 BN was calculated by changing the interlayer spacing and fixing the  $z$ -axis coordinates  
11 of the atoms. When the pressure is 0.026 GPa, the ferroelectric polarization of AB  
12 stacked bilayer BN is  $-1.161 \times 10^{-12}$  C/m, and when the pressure is 0.058 GPa, the  
13 ferroelectric polarization is  $-1.364 \times 10^{-12}$  C/m. Therefore, the ferroelectric polarization  
14 strength of BN can be enhanced by pressure. We also use the MLWF method to  
15 calculate the shape of the  $p_z$  orbital of the N atom under different pressures. The  
16 calculation results show that pressure has little effect on the distortion of the  $p_z$  orbital  
17 of the N atom. The inset in figure. 3(f) shows the ferroelectric switching energy barrier  
18 versus pressure. The ferroelectric sliding barrier increases with increasing pressure.  
19 According to the above conclusions, pressure can not only enhance the ferroelectric  
20 polarization of BN but also stabilize its polarization.  
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#### 38 4. Conclusions

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40 In summary, for 2D AB or BA stacked bilayer BN, we studied the effects of an  
41 external electric field and pressure on ferroelectricity, and the possible origin of the out-  
42 of-plane ferroelectric polarization. Both the external electric field and pressure can  
43 significantly enhance the ferroelectric polarization of BN. In particular, the electric field  
44 can flip the polarization direction of BN. There are both repulsive and attractive  
45 interactions between vertically aligned N atoms and B atoms. The former originates  
46 from N  $p_z$  electrons and B  $p_z$  electrons. The latter originates from N  $p_z$  electrons and B<sup>3+</sup>  
47 ions. These two interactions lead to distortion of the B-top N  $p_z$  orbital. The hexagon-  
48 top N  $p_z$  orbital is also distorted by the electrostatic repulsion of the  $p_z$  orbitals of B and  
49 N atoms on the facing hexagon. These two N  $p_z$  orbital distortions collectively  
50 determine the strength of the ferroelectric polarization. When the electric field is  
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sufficiently large, the paraelectric phase AA becomes stable, which may help the design of logic devices.

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### References

- [1] M. Wu and P. Jena, *WIREs Comput. Mol. Sci.* **8**, e1365 (2018).
- [2] E. Bruyer, D. Di Sante, P. Barone, A. Stroppa, M.-H. Whangbo, and S. Picozzi, *Phys. Rev. B* **94**, 195402 (2016).
- [3] D. Di Sante, A. Stroppa, P. Barone, M.-H. Whangbo, and S. Picozzi, *Phys. Rev. B* **91**, 161401 (2015).
- [4] W. Ding, J. Zhu, Z. Wang, Y. Gao, D. Xiao, Y. Gu, Z. Zhang, and W. Zhu, *Nat. Commun.* **8**, 14956 (2017).
- [5] L. Li and M. Wu, *ACS Nano* **11**, 6382 (2017).
- [6] M. Wu and X. C. Zeng, *Nano Lett* **16**, 3236 (2016).
- [7] M. Wu and X. C. Zeng, *Nano Lett* **17**, 6309 (2017).
- [8] S. N. Shirodkar and U. V. Waghmare, *Phys. Rev. Lett.* **112**, 157601 (2014).
- [9] S. Wan, Y. Li, W. Li, X. Mao, W. Zhu, and H. Zeng, *Nanoscale* **10**, 14885 (2018).
- [10] A. Belianinov, Q. He, A. Dziaugys, P. Maksymovych, E. Eliseev, A. Borisevich, A. Morozovska, J. Banys, Y. Vysochanskii, and S. V. Kalinin, *Nano Lett* **15**, 3808 (2015).
- [11] M. Chyashvichyus, M. A. Susner, A. V. Ievlev, E. A. Eliseev, S. V. Kalinin, N. Balke, A. N. Morozovska, M. A. McGuire, and P. Maksymovych, *Appl. Phys. Lett.* **109**, 172901 (2016).
- [12] B. Xu, H. Xiang, Y. Xia, K. Jiang, X. Wan, J. He, J. Yin, and Z. Liu, *Nanoscale* **9**, 8427 (2017).
- [13] W. Song, R. Fei, and L. Yang, *Phys. Rev. B* **96**, 235420 (2017).
- [14] S. A. Tawfik, J. R. Reimers, C. Stampfl, and M. J. Ford, *J. Phys. Chem. C* **122**, 22675 (2018).
- [15] C. Xiao, F. Wang, S. A. Yang, Y. Lu, Y. Feng, and S. Zhang, *Adv. Funct. Mater.* **28**, 1707383 (2018).
- [16] K. Chang *et al.*, *Science* **353**, 274 (2016).

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3 [17] Y. Bao *et al.*, *Nano Lett* **19**, 5109 (2019).  
4 [18] C. R. Woods, P. Ares, H. Nevison-Andrews, M. J. Holwill, R. Fabregas, F.  
5 Guinea, A. K. Geim, K. S. Novoselov, N. R. Walet, and L. Fumagalli, *Nat.*  
6 *Commun.* **12**, 347 (2021).  
7 [19] K. Yasuda, X. Wang, K. Watanabe, T. Taniguchi, and P. Jarillo-Herrero, *Science*  
8 **372**, 1458 (2021).  
9 [20] M. Vizner Stern, Y. Waschitz, W. Cao, I. Nevo, K. Watanabe, T. Taniguchi, E.  
10 Sela, M. Urbakh, O. Hod, and M. Ben Shalom, *Science* **372**, 1462 (2021).  
11 [21] X. Jiang *et al.*, *Nat. Commun.* **13**, 574 (2022).  
12 [22] A. Bafekry, M. M. Obeid, C. V. Nguyen, M. Ghergherehchi, and M. Bagheri  
13 Tagani, *J. Mater. Chem. A* **8**, 13248 (2020).  
14 [23] A. Bafekry and M. Neek-Amal, *Phys. Rev. B* **101**, 085417 (2020).  
15 [24] M. M. Obeid, *Appl. Surf. Sci.* **508**, 144824 (2020).  
16 [25] A. Bafekry, *Phys. E.* **118**, 113850 (2020).  
17 [26] A. Bafekry, C. Stampfl, B. Akgenc, B. Mortazavi, M. Ghergherehchi, and C. V.  
18 Nguyen, *Phys. Chem. Chem. Phys.* **22**, 6418 (2020).  
19 [27] C. V. Nguyen and N. N. Hieu, *Chem. Phys.* **468**, 9 (2016).  
20 [28] A. Bafekry, C. Stampfl, and F. M. Peeters, *Sci. Rep.* **10**, 213 (2020).  
21 [29] K. D. Pham *et al.*, *Chem. Phys. Lett.* **716**, 155 (2019).  
22 [30] A. Bafekry, M. Ghergherehchi, and S. Farjami Shayesteh, *Phys. Chem. Chem.*  
23 *Phys.* **21**, 10552 (2019).  
24 [31] T. V. Vu, N. V. Hieu, L. T. P. Thao, N. N. Hieu, H. V. Phuc, H. D. Bui, M. Idrees,  
25 B. Amin, L. M. Duc, and C. V. Nguyen, *Phys. Chem. Chem. Phys.* **21**, 22140  
26 (2019).  
27 [32] S. Yin, Q. Luo, D. Wei, G. Guo, X. Sun, Y. Li, Y. Tang, Z. Feng, and X. Dai,  
28 *Phys. E.* **142**, 115258 (2022).  
29 [33] Z. Chang, K. Yuan, Z. Sun, X. Zhang, Y. Gao, G. Qin, and D. Tang, *Phys. Chem.*  
30 *Chem. Phys.* **23**, 13633 (2021).  
31 [34] P. Kumar, B. S. Bhadoria, S. Kumar, S. Bhowmick, Y. S. Chauhan, and A.  
32 Agarwal, *Phys. Rev. B* **93**, 195428 (2016).  
33 [35] E. J. G. Santos and E. Kaxiras, *Nano Lett* **13**, 898 (2013).  
34 [36] F. Ahmed, Y. D. Kim, M. S. Choi, X. Liu, D. Qu, Z. Yang, J. Hu, I. P. Herman,  
35 J. Hone, and W. J. Yoo, *Adv. Funct. Mater.* **27**, 1604025 (2017).  
36 [37] G. Groetzinger, *Nature* **135**, 1001 (1935).  
37 [38] W. H. Huber, L. M. Hernandez, and A. M. Goldman, *Phys. Rev. B* **62**, 8588  
38 (2000).  
39 [39] V. E. Dorgan, A. Behnam, H. J. Conley, K. I. Bolotin, and E. Pop, *Nano Lett* **13**,  
40 4581 (2013).  
41 [40] A. A. Mostofi, J. R. Yates, Y.-S. Lee, I. Souza, D. Vanderbilt, and N. Marzari,  
42 *Comput. Phys. Commun.* **178**, 685 (2008).  
43 [41] N. Marzari, A. A. Mostofi, J. R. Yates, I. Souza, and D. Vanderbilt, *Rev. Mod.*  
44 *Phys.* **84**, 1419 (2012).  
45 [42] J. Qiao, J. Zhou, Z. Yuan, and W. Zhao, *Phys. Rev. B* **98**, 214402 (2018).  
46 [43] M. C. Romano, A. Vellasco-Gomes, and A. Bruno-Alfonso, *J. Opt. Soc. Am. B*  
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- 35**, 826 (2018).
- [44] J. H. Ryoo, C.-H. Park, and I. Souza, Phys. Rev. B **99**, 235113 (2019).
- [45] Z. Fu, B. Yang, and R. Wu, Phys. Rev. Lett. **125**, 156001 (2020).
- [46] G. Kresse and J. Hafner, Phys. Rev. B **47**, 558 (1993).
- [47] G. Kresse and J. Furthmüller, Phys. Rev. B **54**, 11169 (1996).
- [48] J. Hafner, J. Comput. Chem. **29**, 2044 (2008).
- [49] P. E. Blöchl, Phys. Rev. B **50**, 17953 (1994).
- [50] J. P. Perdew, K. Burke, and M. Ernzerhof, Phys. Rev. Lett. **80**, 891 (1998).
- [51] X. Wu, M. C. Vargas, S. Nayak, V. Lotrich, and G. Scoles, J. Chem. Phys. **115**, 8748 (2001).
- [52] G. Henkelman, B. P. Uberuaga, and H. Jónsson, J. Chem. Phys. **113**, 9901 (2000).
- [53] N. Marzari and D. Vanderbilt, Phys. Rev. B **56**, 12847 (1997).
- [54] I. Souza, N. Marzari, and D. Vanderbilt, Phys. Rev. B **65**, 035109 (2001).
- [55] R. M. Ribeiro and N. M. R. Peres, Phys. Rev. B **83**, 235312 (2011).
- [56] G. Constantinescu, A. Kuc, and T. Heine, Phys. Rev. Lett. **111**, 036104 (2013).
- [57] R. S. Pease, Nature **165**, 722 (1950).
- [58] S. Zhou, J. Han, S. Dai, J. Sun, and D. J. Srolovitz, Phys. Rev. B **92**, 155438 (2015).
- [59] J. H. Warner, M. H. Rummeli, A. Bachmatiuk, and B. Büchner, ACS Nano **4**, 1299 (2010).
- [60] C.-J. Kim, L. Brown, M. W. Graham, R. Hovden, R. W. Havener, P. L. McEuen, D. A. Muller, and J. Park, Nano Lett **13**, 5660 (2013).
- [61] W. Tang, E. Sanville, and G. Henkelman, J. Phys.: Condens. Matter **21**, 084204 (2009).
- [62] R. F. J. C. R. Bader, Chem. Rev. **91**, 893 (1991).
- [63] Z. Fu, M. Liu, and Z. Yang, Phys. Rev. B **99**, 205425 (2019).
- [64] Z.-H. Chi, X.-M. Zhao, H. Zhang, A. F. Goncharov, S. S. Lobanov, T. Kagayama, M. Sakata, and X.-J. Chen, Phys. Rev. Lett. **113**, 036802 (2014).
- [65] L. Fu *et al.*, Sci. Adv. **3**, e1700162 (2017).