# Ab initio calculation of the growth of Te nanorods and $Bi_2Te_3$ nanoplatelets<sup>\*</sup>

Tian Xiao-Qing(田晓庆), Du Shi-Xuan(杜世萱), and Gao Hong-Jun(高鸿钧)<sup>†</sup>

Nanoscale Physics & Devices Laboratory, Institute of Physics, Chinese Academy of Sciences, Beijing 100080, China

(Received 16 May 2007; revised manuscript received 6 June 2007)

In this paper the growth mechanism of a Te/Bi<sub>2</sub>Te<sub>3</sub> novel structure is studied by *ab-initio* calculations. The results show that the growth of Te nanorods is determined by the adsorption energy of Te atoms on different crystalline Te surfaces. The adsorption energy of Te on the Te (001) surface is 3.29 eV, which is about 0.25 eV higher than that of Te on the Te (110). This energy difference makes the preferential growth direction along the < 001 > direction. In addition, the higher surface energy of Bi<sub>2</sub>Te<sub>3</sub> (110) and the lattice misfit between crystalline Bi<sub>2</sub>Te<sub>3</sub> and Te along < 001 > direction are considered to explain the growth of the Bi<sub>2</sub>Te<sub>3</sub> nanoplatelets, in which Volmer–Weber model is used. The theoretical results are in agreement with experimental observation.

**Keywords:** *ab initio* calculation, growth mechanism, Te nanorod, Bi<sub>2</sub>Te<sub>3</sub> nanoplatelet **PACC:** 7115A, 6860, 6150

### 1. Introduction

Understanding and controlling the growth of nanostructures and self-assembly structures are of great importance for designing and fabricating nanoscale devices.<sup>[1-4]</sup> Thermoelectric devices are used in many special applications, such as coolers and electrical power generators. The inherent efficiency of thermoelectric material is determined by a dimensionless parameter ZT (the thermoelectric figure of merit), where T is the temperature and Z characterizes the electrical and thermal transport properties of materials. At present, thermoelectric devices are made of bulk material, which have low efficiency at room temperature due to the low ZT. The important properties for high-efficiency thermoelectric materials are large Seebeck coefficient, moderate electrical conductivity, and small thermal conductivity. It has been theoretically predicted and experimentally demonstrated that nanostructure materials exhibit a high value of ZT due to enhanced thermoelectric power and electrical conductivity or reduced thermal conductivity.<sup>[5-12]</sup> A synthesis of novel composite nanostructure of  $Te/Bi_2Te_3^{[13]}$  was reported, in which the Bi<sub>2</sub>Te<sub>3</sub> nanoplatelets grew on Te nanorods by packing them along c-axial direction in a top-bottom sequence. It may be a potential thermoelectric building block due to the confined thickness of the platelet. However, the growth mechanism of  $Te/Bi_2Te_3$  nanostructures has not been reported. In this article, we will present *ab initio* calculations of adsorption energies of Te and Bi adatoms on different surfaces of Te nanorods to explain the growth mechanism. The growth mechanism of the Bi<sub>2</sub>Te<sub>3</sub> nanostructure is also discussed.

### 2. Method

We analyse the adsorption energies of Te adatom on different surfaces, the surface energy of Te (110)and  $Bi_2Te_3(110)$ , and the lattice misfit between  $Bi_2Te_3$  epitaxial layers and Te (110) substrate. We use ab-initio calculations based on density functional theory,<sup>[14]</sup> which is implemented in the Vienna *ab-initio* simulation package.<sup>[15]</sup> Perdew–Burke– Ernzerhof<sup>[16]</sup> generalized gradient approximation for the exchange-correlation energy functional, projector augmented waves method, and 227eV for cut-off energy of the plane waves basis set are employed. First, the crystal structures of the bulk Te and the bulk  $Bi_2Te_3$  are optimized. Second, Te (110), Te (001) and  $Bi_2Te_3$  (110) surfaces are relaxed. Third, for analysing the adsorption energies of Te adatom on different surfaces, all possible adsorption sites for Te on Te (110)

<sup>†</sup>E-mail: hjgao@aphy.iphy.ac.cn

<sup>\*</sup>Project supported by the National Natural Science Foundation of China (Grant No 10774176), the Nation Basic Research Program of China (Grant Nos 2006CB921305 and 2006CB806202).

http://www.iop.org/journals/cpb http://cpb.iphy.ac.cn

and Te (001) surfaces are optimized. The adsorption energy of each adsorption pattern on different surfaces is calculated by

$$E_{\rm adsorption} = E - E_0 - E_{\rm adatom}, \tag{1}$$

where E is the total energy of the surface with one adatom on a possible adsorption site,  $E_0$  is the total energy of the corresponding clean surface, and  $E_{\text{adatom}}$ is energy of Te adatom. Fourth, for analysing the surface energy ( $\sigma$ ) of Te (110) and Bi<sub>2</sub>Te<sub>3</sub>(110), we calculate total energy of Bi<sub>2</sub>Te<sub>3</sub> (110) surface and Te (110) surface, respectively. And then the surface energy is calculated by Boettger equation<sup>[17]</sup>

$$\sigma = [E_{\text{slab}}^N - N\Delta E]/2A, \qquad (2)$$

where

$$\Delta E = (E_{\text{slab}}^N - E_{\text{slab}}^{N-2})/2, \qquad (3)$$

 $E_{\text{slab}}^N$  is the energy of surface with N layers, and A is the surface area.

## 3. Results and discussion

# 3.1. The growth mechanism of Te nanorods

In Ref.[9], the Te nanorods grow along < 001 > direction and the side surface is Te (110) according to transmission electron microscopy, scanning electron microscope and energy dispersive x-ray spectroscopy results. In our calculations, Te nanorods with Te (110), and Te (001) surfaces are considered. Figures 1(a) and 2(a) show all the possible adsorption sites for Te adatom on Te (001) and Te (110) surfaces. Figures 1(d) and 2(d) show the order of height of atomic layers.



**Fig.1.** Adsorption of Te adatom on Te (110). (a) Five typical adsorption sites; (b) Top view of the most stable configuration for Te adsorption; (c) Side view of (b); (d) The red atom stands for the first layer of the slab, argentine for the second layer, and blue for the third layer; the brown atom stands for the Te adatom.

In the case of Te adatom on Te (110) surface, it is found that only bridge site and hollow site are possible adsorption sites. The configuration with Te adatom on the top site of Te (110) is not stable and the adatom moves to bridge site after relaxation. The most stable configuration is that the Te adatom on the bridge1 site, and the adsorption energy is 3.04eV as shown in Figs.1(b) and 1(c), which is 0.62-0.77eV higher than that of other configurations. As to the Te on Te (001) surface, *ab-initio* calculations show that the configurations with Te adatom on the top and bridge sites are not stable and the adatom moves

to the hollow sites after relaxation. The most stable configuration is the one with the Te adatom on the hollow1 site as shown in Figs.2(b) and 2(c), and adsorption energy is 3.29eV, which is 0.22–0.84eV higher than that of other configurations. Therefore, comparing the most stable configuration, we find that the adsorption energy of Te on Te (001) surface is 0.25eV higher than that of Te on Te (110) surface, implying that Te nanorods grow faster along < 001 > direction than along < 110 > direction. It is in good agreement with the experimental result.<sup>[13]</sup>



**Fig.2.** Adsorption of Te adatom on Te (001). (a) Four typical adsorption sites; (b) Top view of the most stable configuration for Te adsorption; (c) Side view of (b) and the lowest energy is in the case of (c), which makes the preferential growth along this direction; (d) Same to Fig.1(d).

# 3.2. The growth mechanism of $Bi_2Te_3$ nanoplatelets

In previous studies, it was found that the  $Bi_2Te_3$ nanoplatelets grew on the side surfaces of Te nanorods along < 110 > direction and the growth of Bi<sub>2</sub>Te<sub>3</sub> is restricted in < 001 > direction.<sup>[13]</sup> First, the Boettger equation is used to calculate the surface energies of both Te (110) and  $Bi_2Te_3$  (110) surfaces. In the case of Te (110), the total energy calculations of the clean surfaces with 5–10 atom layers are carried out respectively. The area of Te (110) surface slab is fixed to 0.465 nm,<sup>2</sup> and the thicknesses of vacuum layers are all set to 10Å. Then the energies of single layer,  $\Delta E$ (calculated by Eq.3), are obtained. They vary from -10.666 to -10.659 eV. Since the relative error is only 0.07%. The mean value of  $\Delta E$  (-10.661eV) is employed to calculate surface energy by using Eq.(2). The surface energy  $\sigma_{\text{Te}(110)}$  is  $1.13 \text{eV/nm}^2$ . In the case of  $Bi_2Te_3$  (110), the surface area is fixed to  $0.816nm^2$ . The thickness of vacuum layer and the number of slab

layers are the same as that used in the Te (110) calculation. The total energies of the surfaces with different layers are calculated. The energy of a single layer,  $\Delta E$  (calculated by Eq.3), varies from -17.794 to -17.786eV. The relative error is only 0.04%. We use the mean value of  $\Delta E$ , -17.790, to calculate surface energy, which is 2.20eV/nm<sup>2</sup>. We find that the surface energy of Bi<sub>2</sub>Te<sub>3</sub> (110) (2.20eV/nm<sup>2</sup>) is nearly twice as much as that of Te (110) (1.13eV/nm<sup>2</sup>).

Second, we analyse the lattice misfit between  $\text{Bi}_2\text{Te}_3$  epitaxial layer and Te (110) substrate. Te has trigonal cell structure with  $a_{\rm s}{=}0.45033$ nm and  $c_{\rm s}{=}0.59614$ nm. Bi<sub>2</sub>Te<sub>3</sub> has hexagonal cell structure with  $a_{\rm e}{=}0.4445$ 9nn,  $c_{\rm e}{=}3.1826$ nm. The  $c_{\rm e}$  is about five times as much as  $c_{\rm s}$ . The lattice misfits between Te crystal and Bi<sub>2</sub>Te<sub>3</sub> crystal are  $f_1 = \frac{a_{\rm s} - a_{\rm e}}{a_{\rm s}} = 0.3\%$ , and  $f_2 = \frac{5c_{\rm s} - c_{\rm e}}{5c_{\rm s}} = -7.36\%$ , where  $f_1$  and  $f_2$  are the lattice misfit of a and c direction between Te and Bi<sub>2</sub>Te<sub>3</sub> crystals respectively. According to the equation  $E_{\varepsilon} = GV f^2/(1-v)^{[18]}$  (where  $E_{\varepsilon}$  is the lattice

tice misfit energy, G is shear modulus, V is the volume of epitaxial island, f is the lattice misfit, and  $\nu$  is Poisson's ratio), the lattice misfit energy is proportional to  $f^2$ . Thus, lattice misfit energy along < 001 > direction is much higher than other directions.



**Fig.3.** Schematic diagram of the growth mode of the  $Bi_2Te_3$  nanoplatelet. The nanoplatelet is grown based on the nanorod as the substrate. This growth is caused by the lattice misfit and difference in surface energy. The growth follows Volmer–Weber model.

Based on the above theoretical consideration, it can be summarized as follows. 1) The surface energy of  $Bi_2Te_3$  (110) is nearly twice as much as that of Te (110). 2) The lattice misfit energy is very large in c direction. Then it can be deduced that the growth of  $Bi_2Te_3$  epitaxial layer on Te (110) substrate follows the Volmer–Weber model described in Ref.[19] (three-dimensional island growth) as shown in Fig.3. The growth of  $Bi_2Te_3$  epitaxial layer is confined along < 001 > direction due to large lattice misfit energy, and that will cause the formation of  $Bi_2Te_3$ nanoplatelets.

### 4. Conclusions

Through *ab-initio* calculations, we find that the Te nanorods growth is preferentially along < 001 >, which is determined by the adsorption energies on different surfaces. Based on the surface energy and lattice misfit, we have also explained the growth of Bi<sub>2</sub>Te<sub>3</sub> nanoplatelets on the Te nanorods. Because of large surface energy of Bi<sub>2</sub>Te<sub>3</sub> (110) and large lattice misfit energy along < 001 > direction, the growth of Bi<sub>2</sub>Te<sub>3</sub> nanoplatelets on Te (110) follows Volmer–Weber model. Thus, Bi<sub>2</sub>Te<sub>3</sub> nanoplatelets along < 001 > direction are obtained. This work is helpful for understanding the formation of Bi<sub>2</sub>Te<sub>3</sub> nanostructures.

# References

[1] Barth J V, Costantini G and Kern K 2005 Nature 437  $_{671}$ 

Du S X, Gao H J, Seidel C, Tsetseris L, Ji W, Kopf H, Chi L F, Fuchs H, Pennycook S J and Pantelides S T 2006 *Phys. Rev. Lett.* **97** 156105

Shi D X, Ji W, Lin X, He X B, Lian J C, Gao L, Cai J M, Lin H, Du S X, Lin F, Seidel C, Chi L F, Hofer W A, Fuchs H and Gao H J 2006 *Phys. Rev. Lett.* **96** 226101

- [2] Iijima S 1991 Nature **354** 56
- [3] Burda C, Chen X B, Narayanan R and EI-Sayed M A 2005 Chem. Rev. 105 1025
  Yang H T, Shen C M, Su Y K, Yang T Z, Gao H J and Wang Y 2003 App. Phys. Lett. 82 4729
  He S T, Xie S S, Yao J N, Gao H J and Pang S J 2002 App. Phys. Lett. 81 150
  He S T, Yao J N, Jiang P, Shi D X, Zhang H X, Xie S S, Pang S J and Gao H J 2001 Langmuir 5 1571
  Liu F, Cao P J, Zhang H R, Li J Q and Gao H J 2004 Nanotechnology 15 949
  [4] Shi D X, Song Y L, Zhang H X, Xie S S, Pang S J and
- Gao H J 2001 Acta Phys. Sin. 50 361
  Xiao C W, Yang H T, Shen C M, Li Z A, Zhang H R, Liu
  F, Yang T Z, Chen S T and Gao H J 2005 Chin. Phys. 14 2269

- [5] Venkatasubramanian R, Sivola E, Colpitts T and Quinn B O 2001 Nature 413 597
- [6] Harman T C, Taylor R J, Walsh M P and La Forge BE 2002 Science 297 2229
- [7] Hsu K F, Guo S L F, Chen W, Dyck J S, Uher C, Hogan T, Polychroniadis E K and Kanatzidis M G 2004 Science 303 818
- [8] Hicks L D and Dresselhaus M S 1993 Phys. Rev. B 47 16631
- [9] Di Salvo F J 1999 Science 285 703
- [10] Balandin A and Wang K L 1998 Phys. Rev. B 58 1544
- [11] Xia Y, Yang P, Sun Y, Wu Y, Mayers B, Gates B, Yin Y, Kim F and Yan H 2003 Adv. Mater. Weinheim Ger. 15 353
- [12] Lee S W and Allmen P V 2006 Appl. Phys. Lett. 88 022107
- [13] Lu W G, Ding Y, Chen Y X, Wang Z L and Fang J Y 2005 J. Am. Chem. Soc. **127** 10112
- [14] Kohn W and Sham L J 1965 Phys. Rev. B 140 A1133
- [15] Kresse G and Furthmiller J 1996 Comput. Matter. Sci. 6 15
- [16] Perdew J P, Chevary J A, Vosko S H, Jackson K A, Pederson M R, Singh D J and Fiolhais C 1992 Phys. Rev. B 46 6671
- [17] Boettger J C 1994 Phys. Rev. B 49 16798
- [18] Matthews J M 1979 Dislocations in Solids vol 2 (Amsterdam: North-Holland) p471
- [19] Zhang Z Y and Lagally M G 1997 Science 276 377